Nuclear Chemistry

Nuclear chemistry is the study of atomic nuclei and the changes they undergo. It plays a critical role in fields such as medicine, energy production, and environmental research.







Introduction

- Definition of nuclear chemistry
- Importance of nuclear chemistry in various fields



Nuclear Reactions

Basics of Nuclear Reactions

Nuclear reactions involve changes in atomic nuclei, including the formation or decay of isotopes.

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Different Types of Nuclear Reactions

Examples include nuclear fission, nuclear fusion, and radioactive decay.



Nuclear Decay

Radioactive Decay

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Radioactive decay is the spontaneous breakdown of atomic nuclei, emitting radiation in the process.

Types of Radioactive Decay

Common types include alpha decay, beta decay, and gamma decay.



😆 Made with Gamma

Half-Life

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Definition of Half-Life

Half-life is the time it takes for half of a radioactive substance to decay.

2 Calculation of Half-Life

Half-life can be calculated using the decay constant and the initial amount of a radioactive substance.



Nuclear Energy

1 Nuclear Fission

Nuclear fission is a process where the nucleus of an atom is split, releasing a large amount of energy. **Nuclear Fusion**

2

Nuclear fusion is a process where two atomic nuclei combine to

