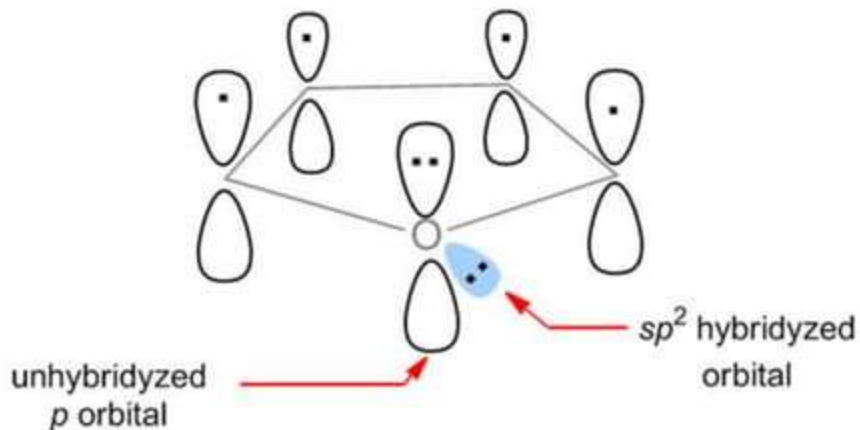


FURAN

Properties

1. Aromaticity



Properties

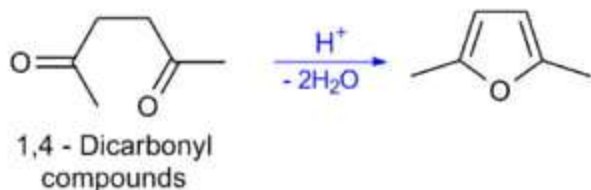
1. Aromaticity

- Furan have 4 C and 1 O , all are sp^2 hybridized
- sp^2 hybridization is **planar**, it makes a planar furan ring structure.
- Each ring atom also contains unhybridized p orbital that is perpendicular to the plane of σ bonds (plane of ring).
- Here p orbitals are parallel to each other, so overlapping btwn p orbitals is possible.
- the total nu of non bonding e- are 6 (4 of four C, 2 from one O)
- The resonance of 6 e- follows the Hückel's rule
- So the furan is aromatic .

Synthesis

1. Paal-Knorr synthesis of furan

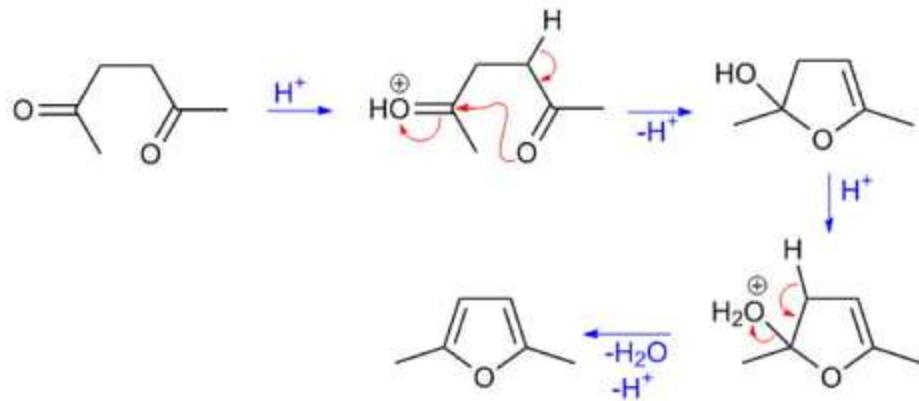
- Acid catalysed ,cyclising dehydration of 1,4 - dicarbonyl compounds.



Synthesis

1. Paal-Knorr synthesis of furan

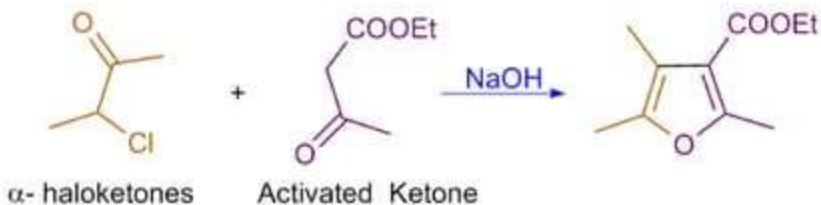
Mechanism



Synthesis

2. Feist – Benary Synthesis

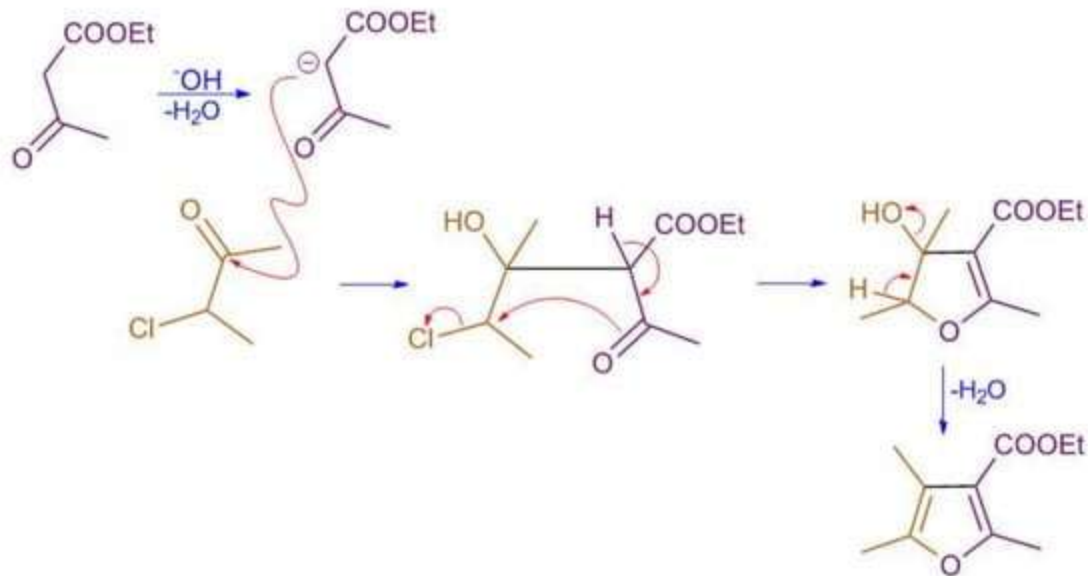
- Reaction of α - haloketones with β -ketoesters in the presence of a base (not ammonia) to give furans.



Synthesis

2. Feist – Benary Synthesis

Mechanism

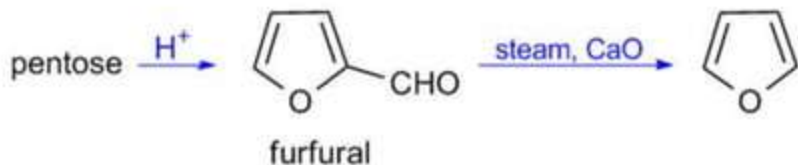


Synthesis

3. From carbohydrate

Step_1: distillation of CH with sulfuric acid

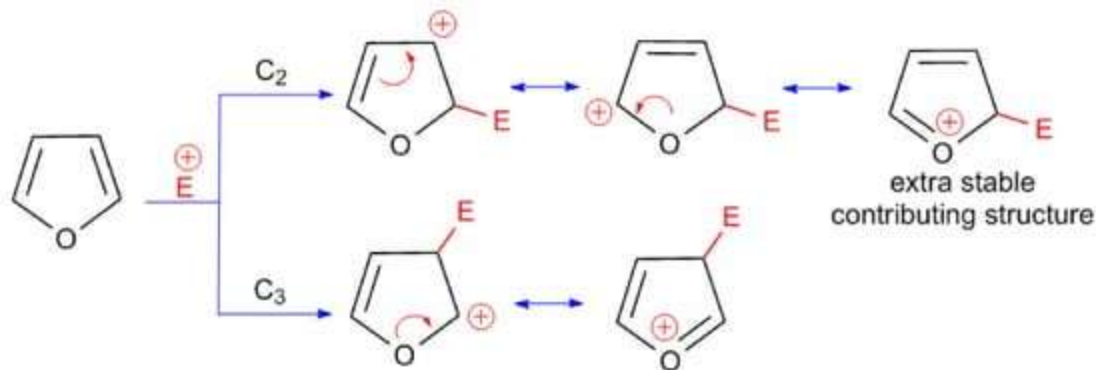
Step_2: catalytic decomposition of furfural in steam



Reactions

1. Electrophilic substitution

furan undergoes electrophilic substitution reaction at 2nd position

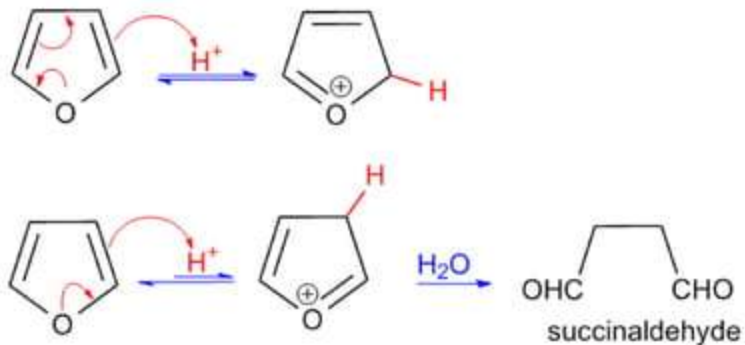


2 reasons...

- C₂ attack gives more resonance contributing structures than C₃.
- Extra stable contributing structure generates upon C₂ attack

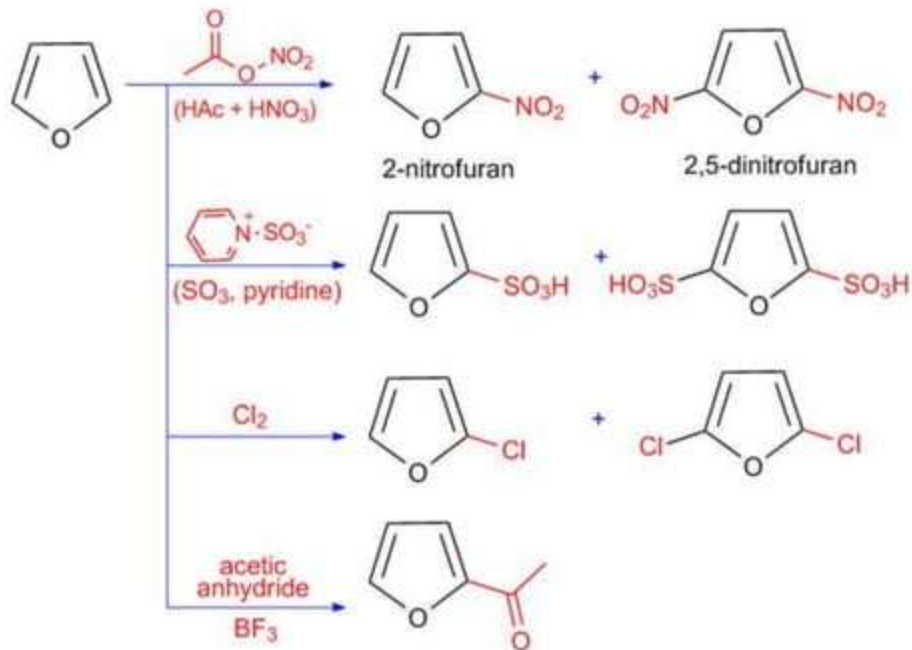
Reactions

1. Electrophilic substitution



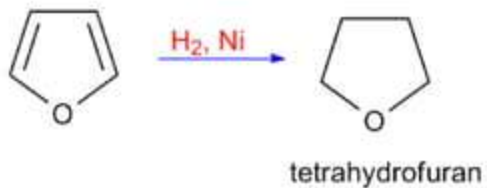
Reactions

1. Electrophilic substitution



Reactions

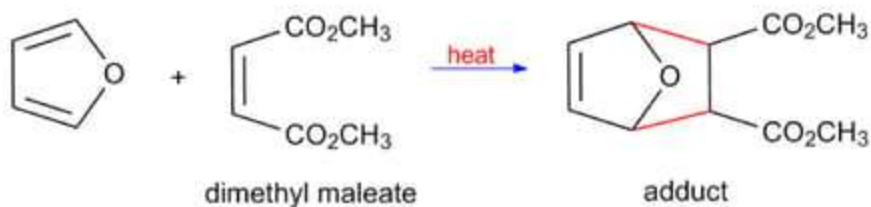
2. Reduction



Reactions

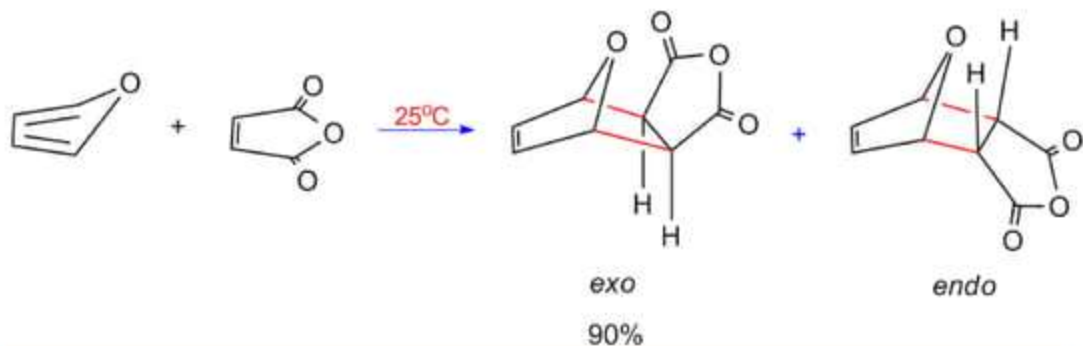
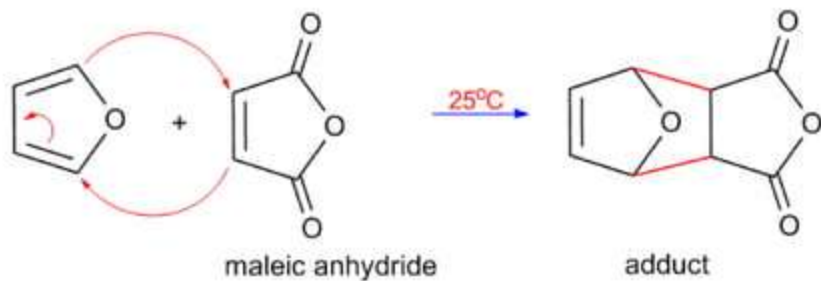
3. Diels-Alder reaction

- O atom is highly electronegative, so the delocalization of lone pair e⁻ (in aromatic system of furan) is not overly effective.
- Thus...Furan can behave as a dienophile and gives 4 + 2 cycloaddition.



Reactions

3. Diels-Alder reaction



Reactions

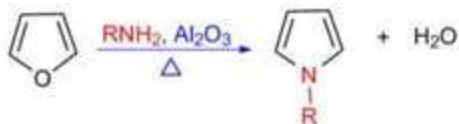
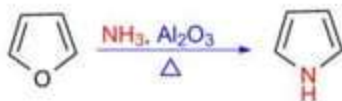
4. Pyrrole synthesis

PYRROLE

Synthesis

1. From Furans

- Industrial process
- Passing furan over **ammonia** in presence of **alumina** as catalyst at high temp.



Medicinal uses

- (1) Nitrofurans: *Nitrofurazone*, *Furazolidone*, *Nitrofurantoin* , Anti-infective Agents: used as an antiprotozoal agent to treat trypanosomiasis and leishmaniasis



- (2) *Ranitidine*: Antihistaminic Agents, used in peptic ulcer

