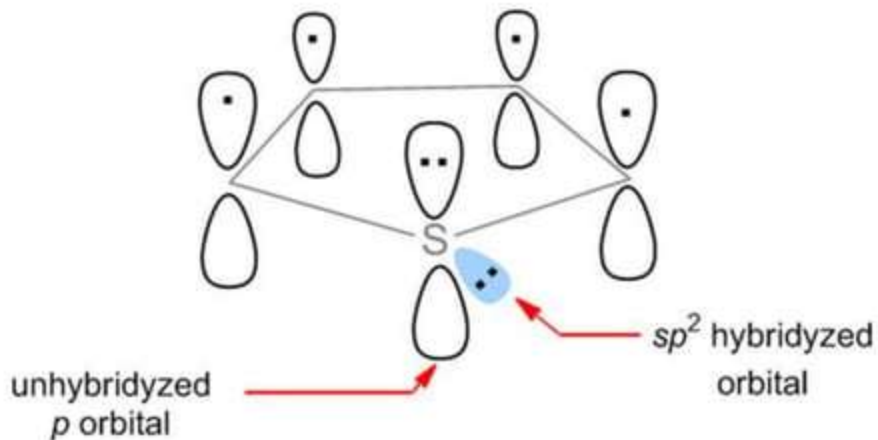
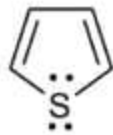


THIOPHENE

Properties

1. Aromaticity



Properties

1. Aromaticity

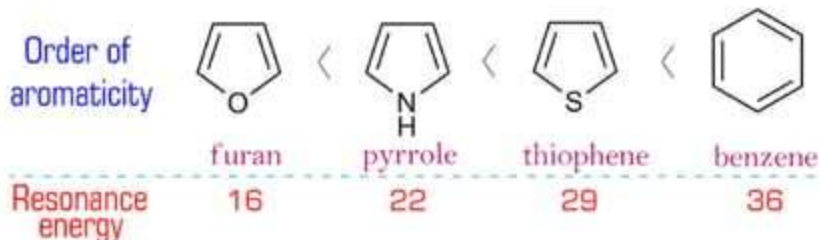
- Thiophene have 4 C and 1 S , all are sp^2 hybridized
- sp^2 hybridization is **planar**, it makes a planar thiophene ring structure.
- Each ring atom also contains unhybridized p orbital that is perpendicular to the plane of σ bonds (plane of ring).
- Here p orbitals are parallel to each other, so overlapping btwn p orbitals is possible.
- the total nu of non bonding e- are 6 (4 of four C, 2 from one S)
- The resonance of 6 e- follows the Hückel's rule
- So the thiophene is aromatic .

THIOPHENE

Properties

1. Aromaticity

Furan is less aromatic / thiophene is more aromatic

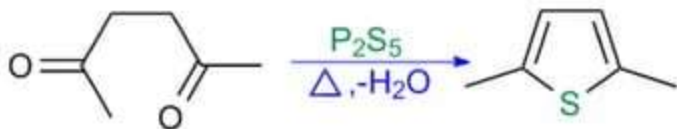


- This order depends on **order of electronegativity** of heteroatoms
- The more e-ve is the atom \rightarrow the more tightly holds its lone pair of e-
 \rightarrow more reduce the *ease of delocalization (aromaticity)*.
- So most e-ve O in furan most decrease aromaticity.
& Least e-ve S in thiophene least decrease aromaticity.
- Thus

Synthesis

1. Paal-Knorr synthesis of thiophene

- The condensation of 1,4-dicarbonyl compounds with sulfur sources gives thiophene.

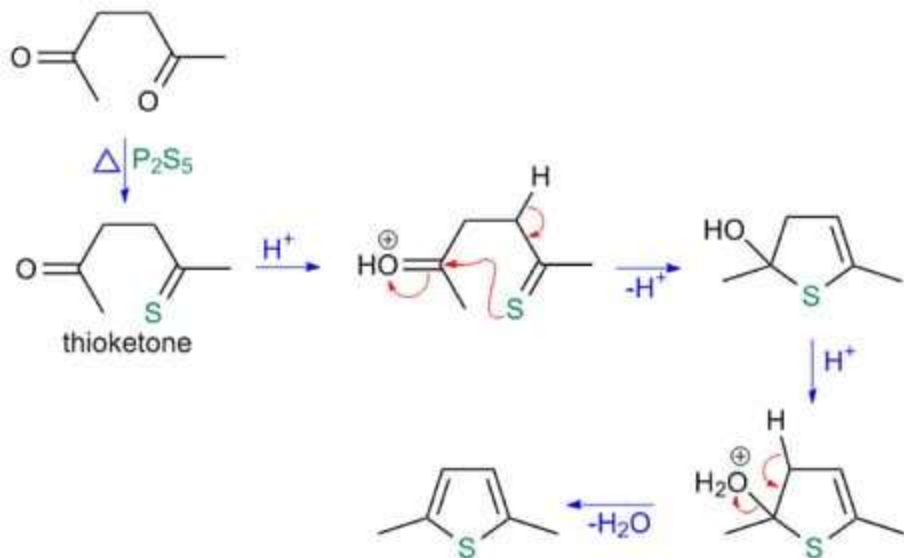


THIOPHENE

Synthesis

1. Paal-Knorr synthesis of furan

Mechanism

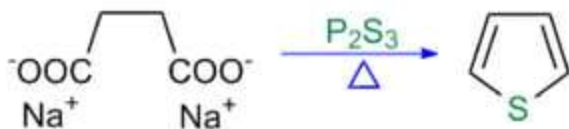


THIOPHENE

Synthesis

2. From sod. succinate

- Laboratory synthesis
- Heating a mix. of sod. succinate and phosphorus trisulfide.

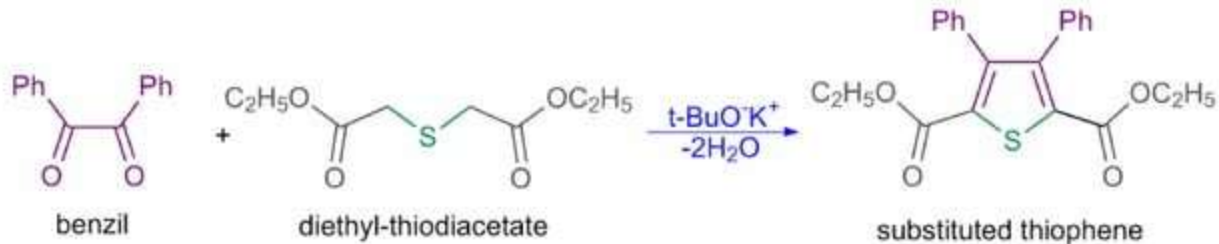
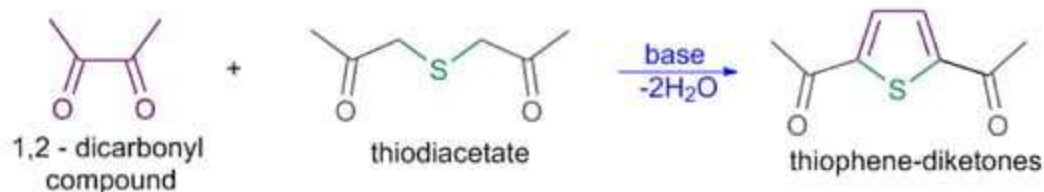


THIOPHENE

Synthesis

3. Hinsberg Synthesis

- Condensation between a 1,2 - dicarbonyl compound and diethyl thiodiacetate in presence of strong base give thiophene 2,5 - diacids (- diketone)

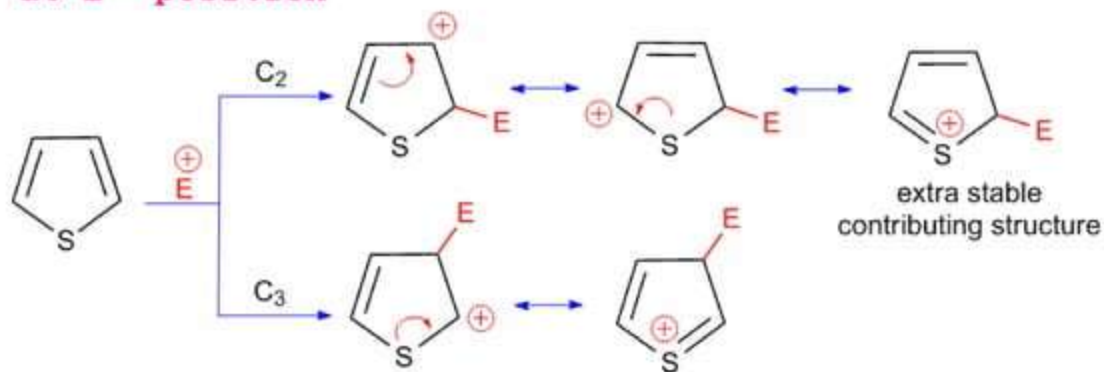


THIOPHENE

Reactions

1. Electrophilic substitution

thiophene undergoes electrophilic substitution reaction at 2nd position



2 reasons...

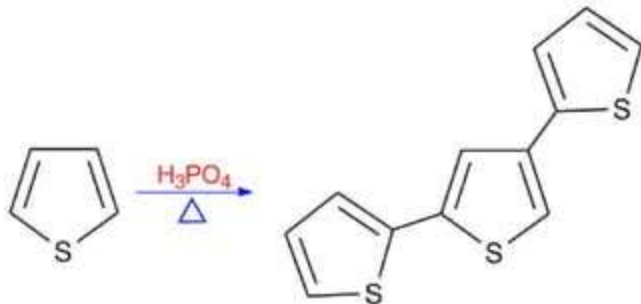
- C₂ attack gives more resonance contributing structures than C₃.
- Extra stable contributing structure generates upon C₂ attack

THIOPHENE

Reactions

1. Electrophilic substitution

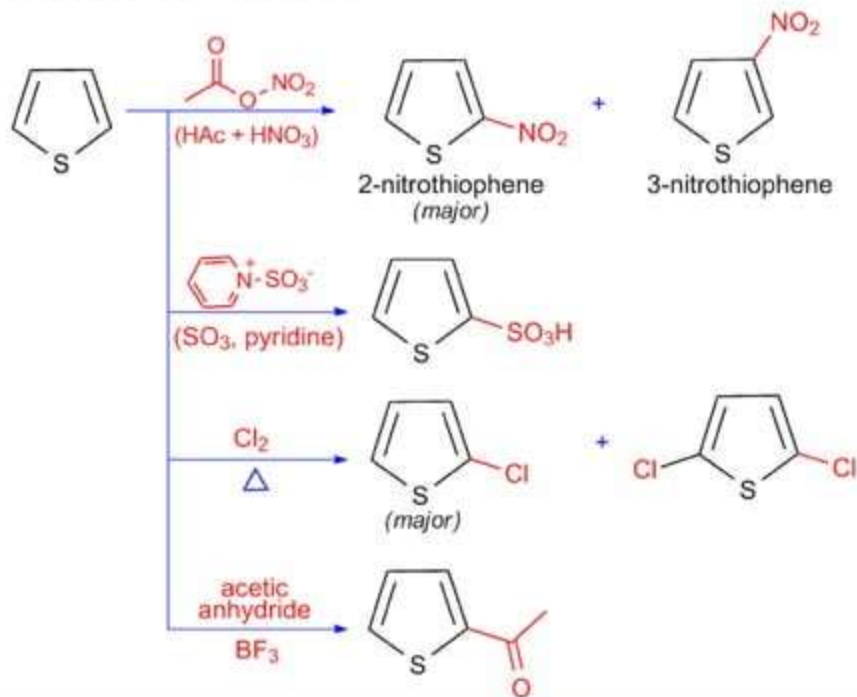
- Very **strongly acidic conditions** lead to acid – catalysed polymerization.
- The action of hot phosphoric acid on thiophene leads to a trimer.



THIOPHENE

Reactions

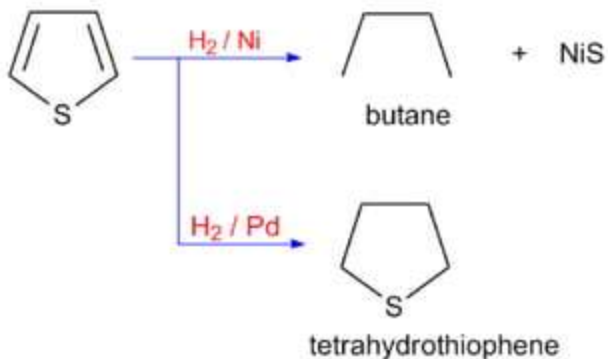
1. Electrophilic substitution



THIOPHENE

Reactions

2. Reduction

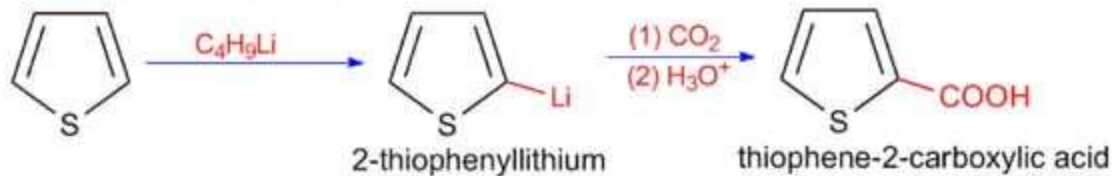


THIOPHENE

Reactions

3. reaction

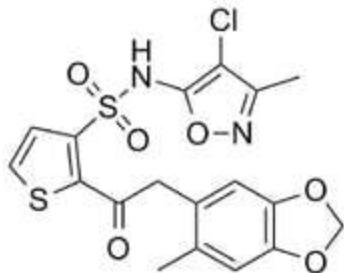
reaction with organolithium



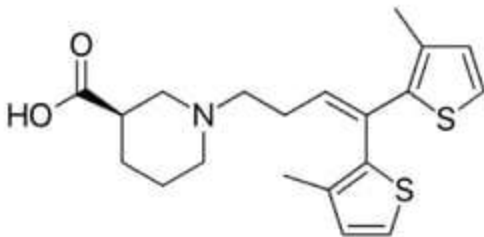
THIOPHENE

Medicinal uses

- (1) *Sitaxsentan*: Cardiovascular Agent, used in pulmonary artery hypertension



- (2) *Tiagabine*: Anticonvulsant Agent, used in the treatment of epilepsy



THIOPHENE

Medicinal uses

(3) *Articaine*: Anesthetic Agent

